## New Ni(II) complexes with 2-aminomethylbenzimidazole, preparation and characterization

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An important class of heterocyclic compounds in many natural and synthetic compounds for the development of new drugs are benzimidazoles [1]. Use of benzimidazoles can be found in liquid crystal materials due to their special physicochemical properties based on fluorescence modulation mechanisms [2]. The benzimidazole moiety involves in a variety of biological processes. For example, N-ribosyl-dimethylbenzimidazole is part of the chemical structure of vitamin  $B_{12}$  [3]. Benzimidazole derivatives are of intensive researches due to their coordination ability besides their biological importance [4].

New Ni(II) complexes of general formula NiL<sub>*x*</sub>Y<sub>*z*</sub> have been synthesized, where L is 2aminomethylbenzimidazole and Y are inorganic anions such as Cl<sup>-</sup>, Br<sup>-</sup>, ClO<sub>3</sub><sup>-</sup>, ClO<sub>4</sub><sup>-</sup>, NO<sub>2</sub><sup>-</sup>, N<sub>3</sub><sup>-</sup>, SCN<sup>-</sup>, SO<sub>4</sub><sup>2-</sup> as well as organic anion CH<sub>3</sub>COO<sup>-</sup>. All newly prepared complex compounds were characterized by X-ray structural analysis and by spectral techniques such as infrared spectroscopy and UV-VIS spectroscopy. Magnetic measurements were also performed for some compounds.

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