

# Study of dicarboxylatecobalt(II) complexes with succinic acid

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This thesis deals with the study of spectral (infrared and electronic) and magnetic properties and the molecular and crystal structures of cobalt complexes with succinic acid. Reaction of  $\text{Co}^{II}$  salt,  $\text{Na}_2\text{suc}\cdot 6\text{H}_2\text{O}$  and corresponding benzimidazole derivative (1-H-benzimidazole - bzim, 2-methyl-1-H-benzimidazole - 2-mebzim, 2-ethylbenzimidazole - 2-etbzim, 2-hydroxy-methyl-1-H-benzimidazole - 2- $\text{CH}_2\text{OHbzim}$ ) or nicotinamide (nia) in a mixed solvent water/methanol (1:1) were prepared six new coordination compounds,  $[\text{Co}(\mu_2\text{-suc})(\text{bzim})_2(\text{H}_2\text{O})_2]_n$  (**1**),  $[\text{Co}(\mu_2\text{-suc})(\text{bzim})_2(\text{MeOH})_2]_n$  (**2**),  $[\text{Co}(\mu_2\text{-suc})(2\text{-mebzim})_2(\text{H}_2\text{O})_2]_n$  (**3**),  $[\text{Co}(2\text{-CH}_2\text{OHbzim})_2(\text{H}_2\text{O})_2]\cdot\text{suc}$  (**4**),  $[\text{Co}(\mu_2\text{-suc})(2\text{-etbzim})_2]_n$  (**5**) and  $[\text{Co}(\mu_2\text{-suc})(\text{nia})_2(\text{H}_2\text{O})_2]\cdot 2\text{H}_2\text{O}_n$  (**6**) of a different composition and molecular structure. Obtained coordination polymers were characterized by elemental analysis, mass spectroscopy, IR, Raman and electron spectroscopy and X-ray structural analysis. Dianions  $\text{suc}^{2-}$  are coordinated to central ion  $\text{Co}^{II}$  in bis(monodentate) bridging manner, leading thus to formation of polymeric chain.

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