

## Study of the Niobium Phosphate glasses

Petr Hejda<sup>1</sup>, Holubová Jana<sup>1</sup>, Černošek Zdeněk<sup>1</sup>, and Černošková Eva<sup>2</sup>

<sup>1</sup>*University of Pardubice, 532 10 Pardubice, Studentská 95, 530 09, Pardubice, Czech Republic*

<sup>2</sup>*Joint Laboratory of Solid State Chemistry of Institute of Macromolecular Chemistry of Czech Academy of Sciences, v.v.i., and University of Pardubice, Pardubice, Czech Republic.*

The phosphate glasses are intensively studied through several decades because of their potential technological applications due to some unique properties, i.e. low synthesis temperature, UV transmission or higher thermal expansion coefficient. They can be used in laser technologies, sealing waxes, as bioimplantates, composites and also storage of nuclear waste [1].

Binary phosphate glasses have relatively limited interval of chemical stability, because ultraphosphate glasses are hygroscopic and polyphosphate glasses can be prepared only in interval  $\sim 35\text{-}50$  mol

Niobium phosphate glasses were previously investigated [2] to obtain chemical resistant and thermal stable materials. However, in all previously studied compositions, PbO was added as a glass modifier; these glasses were very stable, and chemically resistant. The bridging oxygen is now bonded to Nb<sup>5+</sup>, forming O-P-O-Nb-O- type chains. It was also shown that P-O type bonds are mostly found in terminal sites of the chain, while Nb-O bonds are located in the middle of the chains [3].

In this work were studied the 17 glass samples in four compositional lines - 1)  $x\text{Nb}_2\text{O}_5\text{-}(50\text{-}x)\text{ZnO-}50\text{P}_2\text{O}_5$  ( $0 \leq x \leq 15$  mol

[1] R. K. Brow, J. Non-Cryst. Solids 263&264 (2000) 1–2.

[2] F.F. Sene, J.R. et al., J. of Non-Cryst. Solids 348 (2004) 30–37.

[3] A. El Jazouli, et al., J. Solid State Chem. 73 (1988) 433.